

# Sydnones: A Brief Review

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**Abstract:** Sydnones are mesoionic heterocyclic aromatic compounds. They have been widely studied for some important biological activities like antiviral, antitumor, antimicrobial, anti-inflammatory, anticancer, analgesic, anthelmintic and antihypertensive activities. The aim of the present article is to review the available information on sydnones and the derivatives of sydnones and also a look at the future perspectives.

**Keywords:** Sydnone, mesoionic, oxadiazolium.

## INTRODUCTION

Sydnones are unique, dipolar heteroaromatic members of the general class of mesoionic compounds. Chemically they are 1, 2, 3-oxadiazolium-5-olates [1]. A large number of sydnone derivatives have been synthesized with biological interest [2-4] and reported to possess a wide spectrum of biological activities such as antiviral [5], antimicrobial [6], anti-inflammatory [6], analgesic [6], anthelmintic [6], antitumor [7], free radical scavenging [8] and nitric oxide donor [9] activity. The potential value of sydnones as biologically active substances is found in their planar aromatic character, their relatively small size and variation in electron density around the ring. It is believed that the ionic resonance structures of sydnones promote significant interactions with biological molecules. The stimulant drugs feprosidine and mesocarb are substructures of sydnones imine in which the keto group of sydnones (=O) is replaced by imino group (=NH).

## PHYSICAL PROPERTIES

With a few exceptions, sydnones are stable compounds that exhibit considerable polarity. Arylsydnones are generally crystalline solids. Alkylsydnones are often liquids or low-melting solids that can be distilled *in vacuo* without appreciable decomposition. Concentrated acids cause degradation of sydnones yielding hydrazine derivatives [10]. Sydnones are soluble in a variety of organic solvents with the main exceptions being nonpolar solvents such as petroleum ether and hexanes. Additionally, sydnones are generally not water-soluble unless a polar functional group is present.

## CHEMISTRY

### Synthesis

Since their original preparation [11] in 1935 by Earl and Mackney, the only general route to sydnones (2) is still *via* the cyclodehydration of *N*-substituted *N*-nitroso- $\alpha$ -amino

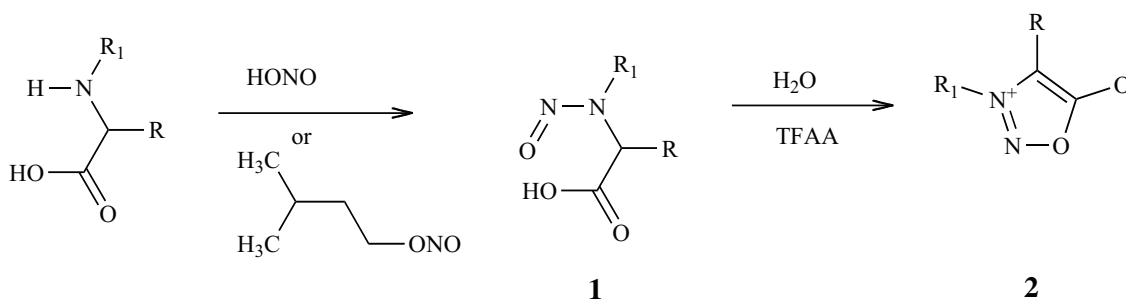
acids as shown in (Scheme 1). While the substituent R can be alkyl, aryl, or hydrogen, the R<sup>1</sup> must be alkyl or aryl. If R<sup>1</sup> is a hydrogen, prototropy occurs to afford a neutral species. The preparation of the *N*-nitroso amino acid generally involves the nitrosation of an *N*-substituted glycine with nitrous acid. Since this nitrosation step requires the use of strongly acidic reaction conditions, sydnones containing acid-sensitive functional groups have been unattainable by this methodology. However, a variation upon this standard method has been developed in which nitrosation is effected under neutral conditions using isoamyl nitrite and dimethoxyethane [12]. Thus, some otherwise unattainable sydnones can now be successfully generated in good to excellent yield.

Earl and Mackney originally employed acetic anhydride at room temperature for six days for cyclodehydration[11]. Since then, several modifications have been made that include heating in acetic anhydride or thionyl chloride, treatment with phosphorus pentoxide, or the use of trifluoroacetic anhydride (TFAA). The reaction with TFAA has become the method of choice since it usually occurs rapidly (<15 minutes) at low temperature (-5°C to 0°C) and in high yields (>90% for *N*-phenylsydnone). The only foreseeable drawback to its use is the far greater cost of this reagent compared to others.

Newer synthetic strategies for accomplishing the aforementioned cyclization have also been put forth. Amongst these are the uses of acetic anhydride at room temperature facilitated by ultrasonification [13], halo iminiumsalts [14], *N,N*-dimethylchlorosulfitemethaniumchloride [15] and 2-chloro-1,3-dimethylimidazolium chloride [16]. Although these new methods are interesting, it is unlikely that they will replace the fast, efficient, and reliable TFAA cyclization.

Azarifar *et al.*, have reported several one-pot syntheses of sydnones [17]. One approach employs dibromo-dimethylhydantoin [18]. This one-pot procedure avoids isolation of the toxic nitrosamine intermediate and makes use of cheap commercially available materials. Moreover, this approach gives good yields for the preparation of a range of sydnones among all methods. Synthesis of sydnones from

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Scheme (1).

readily available anilines have also been carried out to obtain fused ring sydrones [19].

### SPECTRAL STUDIES

In their NMR spectra, the proton (when present) at the C-4 position of the sydnone ring is greatly deshielded in comparison to saturated congeners, usually appearing between 6.5-7.5 ppm. This, then, suggests a polar nature and the presence of an aromatic ring current. Additionally, the infrared spectra of sydrones include two very prominent features, namely a strong carbonyl stretch at  $\sim 1730\text{-}1760\text{ cm}^{-1}$  and a stretch of medium intensity at  $\sim 3150\text{ cm}^{-1}$  for the C-H absorption of the C-4 ring proton (when present) [20]. Moreover, the latter is different from what would be expected for either an alkyl or aryl substituent or from an epoxide with comparable ring strain, which absorb at around  $2900\text{-}3050\text{ cm}^{-1}$ , and therefore is particularly useful in determining if the C-4 position in sydrones are substituted or not.

With regard to the carbonyl stretch, a single, strong band is usually observed. However, due to Fermi resonance splitting, in some cases multiple bands have been observed [21]. Further, as stated above, the sydnone carbonyl typically appears at  $\sim 1730\text{-}1760\text{ cm}^{-1}$  but, when compared to congeneric carbonyl containing species, such as a  $\gamma$ -lactone [which absorbs at  $1770\text{ cm}^{-1}$ ] and tropone [which absorbs at  $1638\text{ cm}^{-1}$ ], one might conclude that the exocyclic C=O bond at the sydnone C-5 position is closer in length to that of a double bond than a single bond. This contention is further supported by the results of X-ray crystallographic analysis of various 3-substituted and 3,4-disubstituted sydrones which show that the exocyclic C=O bond is closer in length to that of a double bond. However, integrated absorption measurements suggest that a high degree of carbonyl bond polarization, rather than bond strength, is responsible for the relatively high energy of absorption. Additionally, molecular orbital calculations and vibrational force constants obtained from vibrational spectra indicate a lower  $\pi$ -bond order for the sydnone carbonyl than those of alicyclic esters [22], thus supporting the argument that contributions from other vibrational modes cause the sydnone carbonyl group to absorb at a higher frequency than anticipated.

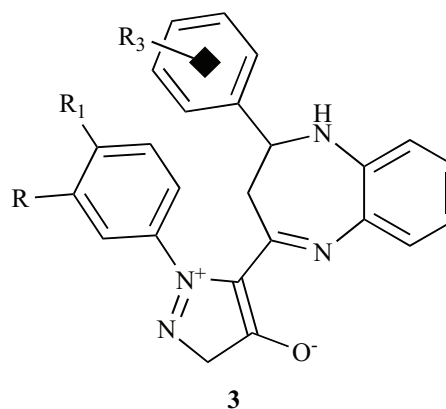
It has been demonstrated, by both theoretical and spectroscopic studies, that protonation of a sydnone moiety occurs at the exocyclic oxygen [23]. This complements earlier investigations where bond orders and charge densities

of various sydrones were correlated to the calculated and observed dipole moments and the observed UV maxima [24]. These studies thus support the contention that substantial charge density resides on the exocyclic oxygen.

### PHARMACOLOGICAL USES

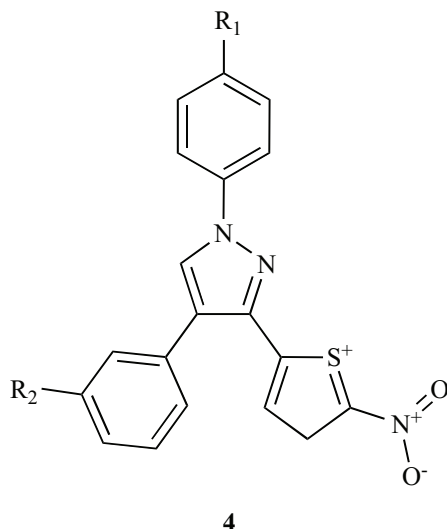
#### Antimicrobial

Prashanth Shinge *et al.*, have synthesized a series of benzodiazepine derivatives of sydrones (3) and screened them for their *in vitro* antimicrobial activity against two bacteria, namely *E. coli* and *C. bacillus* and two fungal cultures, *A. candida* and *R. bataticola* [25,26] using reference drugs Norfloxacin and Griseofulvin, respectively. The same investigators have carried out a one-pot synthesis of 4-(4-chlorophenylazo)-5-methyl-2-aryl-1, 2-dihydropyrazol-3-ones from 3-arylsydrones and studied them for their antimicrobial activity. All the synthesized compounds exhibited more antimicrobial activity than the standard drugs. The results of the antibacterial screening show that compounds substituted with halogen, methyl and nitro exhibit growth inhibitory activity that are more relevant than that of the reference compounds and this activity varies with the substitutions at the phenyl rings.

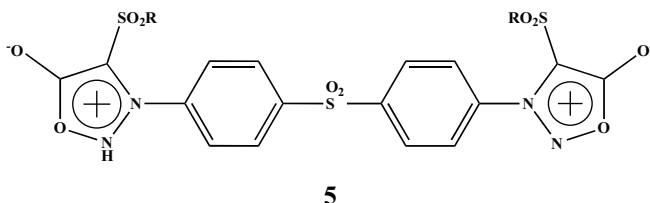


Satheesha Rai *et al.*, have synthesized 1,3,4-trisubstituted pyrazoles(4) via 1,3-dipolar cycloaddition of sydrones and studied their antibacterial activity against four bacterial strains, namely *E.coli* (ATCC-25922), *Staphylococcus aureus* (ATCC-25923), *Pseudomonas aeruginosa* (ATCC-27853) and *Bacillus subtilis* (recultured) [27]. The antifungal activity was studied against the fungi *Candida albicans* (NCIM No. 3100), by the serial dilution method. The tested compounds showed good inhibition at a concentration of

0.125 mg/mL concentration. They also studied the antimicrobial activity of sydnones sulphonamide compounds and found them to be effective antimicrobial agents. The studies revealed that the presence of cholrine atom in the aryl moiety increased the potency of the pyrazoles against the tested bacterial strains.



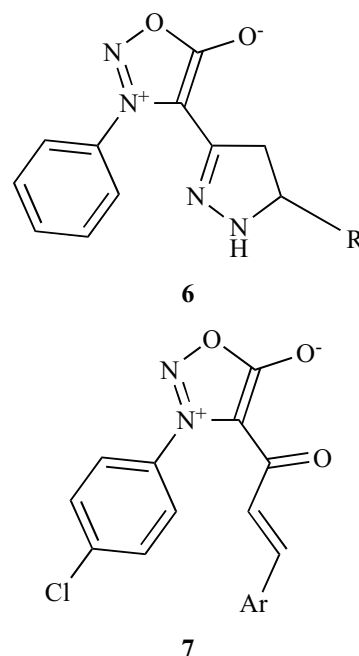
Sharukh Asundaria *et al.*, have studied the antimicrobial activity of a set of bis-sydnones (5) against gram positive bacteria, namely *S.pneumoniae* and *S.aureus* and gram negative bacteria *E.coli* and *P.aeruginosa* [28,29]. The reference drugs used were Streptomycin and Penicillin-G, respectively. The compounds synthesized exhibited moderate activity when compared to the standard.



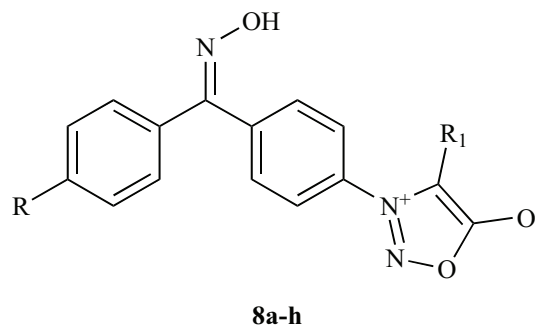
### Antiinflammatory

Satyanarayana *et al.*, have synthesized 4-[5-(substituted aryl)-4,5-dihydro-1H-pyrazol-3-yl]-3-phenylsydnones (6) by condensing 4-[1-oxo-3-(substituted aryl)-2-propenyl]-3-phenylsydnones with hydrazine to yield fourteen compounds [30]. The compounds were tested for their antiinflammatory activity with phenylbutazone and ibuprofen as standards. The halogenated derivatives were found to exhibit more activity compared to other synthesized compounds. These when further tested for their antiarthritic activity showed activity equal to or less than the standard.

Shreenivas Deshpande *et al.*, have studied the antiinflammatory activity of 4-[1-oxo-3-(substituted aryl)-2-propenyl]-3-(4-chlorophenyl) sydnones (7) using carageenan induced rat paw edema method [31]. The synthesized compounds were found to have significant antiinflammatory activity comparable to the standard, namely Ibuprofen.



Ravindra Kamble *et al.*, synthesized and studied benzophenone oxime derivatives of sydnones (8a-h) for their antiinflammatory activity [32]. The synthesized compounds were also evaluated for their inhibitory activity against purified phospholipase A<sub>2</sub> enzymes from snake venom and human inflammatory pleural and ascites fluid. The compounds exhibited activity similar to that of the standard Luffariellin B when a hydrophobic group is present in the aroyl moiety and the 4<sup>th</sup> position of the sydnones ring was substituted by a phenyl ring. It was also observed that a hydrophobic alkyl group reduced the activity whereas the absence of either group resulted in very weak activity.



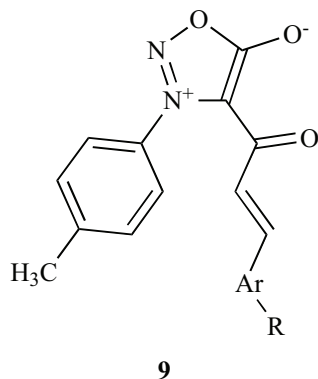
a; R = H, R<sub>1</sub> = H, b; R = CH<sub>3</sub>, R<sub>1</sub> = H, c; R = n-propyl, R<sub>1</sub> = H, d; R = n-butyl, R<sub>1</sub> = H,

e; R = H, R<sub>1</sub> = Ph, f; R = CH<sub>3</sub>, R<sub>1</sub> = Ph, g; R = n-propyl, R<sub>1</sub> = Ph, h; R = n-butyl, R<sub>1</sub> = Ph.

### Analgesic

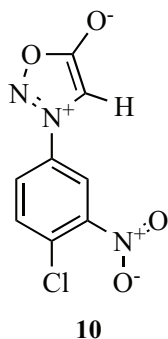
Satyanarayana *et al.*, have synthesized and evaluated 4-[5-(substituted aryl)-4,5-dihydro-1H-pyrazol-3-yl]-3-phenylsydnones and their derivatives for analgesic activity at 100mg/kg in acetic acid induced assay in mice [30]. Among the six compounds that exhibited significant activity, 4-fluoro compound exhibited the highest activity in the series. The standard used was aspirin. These same authors have

synthesized and evaluated some 4-[1-Oxo-3-(substituted aryl)-2-propenyl]-3-(4-methylphenyl) sydrones (9) [33]. The compounds having electron-withdrawing substituents like chloro and nitro at *para* position exhibited more antibacterial activity. The furyl and *para* chloro phenyl analogs showed good analgesic activity. Thus 3-(substituted phenyl)sydnone derivatives possessing substituted styrylketone moieties appear to be another interesting source of antibacterial and analgesic compounds.



### Anticancer

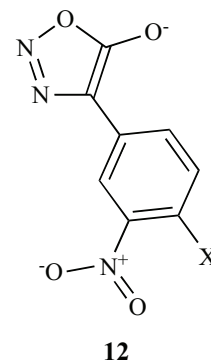
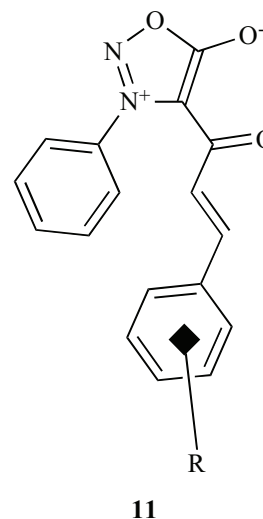
Gerusa Clazer Halila *et al.*, have studied the antitumor activity of SYD-1(3-[4-chloro-3-nitrophenyl]-1,2,3-oxadiazolium-5-olate)(10) [34]. They showed that the effect of this compound on mitochondrial metabolism is due to its inhibitory effect of electron transport through the respiratory chain.



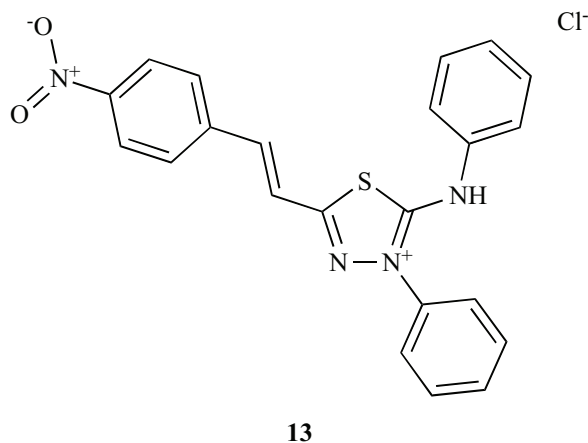
Satyanarayana *et al.*, have synthesized a novel structural feature containing 4-[1-oxo-(substituted aryl)-2-propenyl]-3-phenylsydrones (11) and screened them for their anticancer activity [35]. These compounds contained two pharmacophores, an unsaturated ketone moiety and a sydnone ring. All the three compounds synthesized exhibited promising *in vitro* cytotoxicity in 56 cell lines representing cancers of non-small cell lung, colon, CNS, melanoma, ovarian, prostate, breast and leukemia. The average growth inhibition of 50% was in the range of 1.7-3.5. The methyl derivative was highly selective against SNB-75 tumor cell line of CNS. It was active at less than one nano mole. However, it showed moderate *in vivo* activity in hollow fiber assay model.

Grynberg *et al.*, have studied the effect of some aryl sydrones on murine tumors [36]. The compounds, 3-[4-X-3-nitrophenyl]-1,2,3-oxadiazolium-5-olates (12), where X =

Cl, pyrrolidino, piperidino and morpholino, were synthesized and studied for the survival of mice bearing Sarcoma 180, Ehrlich carcinoma, B10MCII (Fibrous histiocytoma) and L1210 leukemia ascitic tumours proliferation of cultured tumour cells and on the synthesis of DNA in L1210 leukemia. The chloro and pyrrolidino derivatives significantly enhanced the survival of S180, Ehrlich and B10MCII tumour-bearing mice in *in vivo* studies. Further the pyrrolidino derivative showed significant activity against L1210. Piperidino and morpholino derivatives did not show any antitumour activity. The chloro derivative was the most cytotoxic against all the above tumour cells. All the compounds tested inhibited thymidine uptake by L1210 cells, the morpholino derivative being the least active.



Senff Ribeiro *et al.*, have studied the effect of 4-Phenyl-5-(4-nitro-cinnamoyl)-1,3,4-thiadiazolium-2-phenylamine chloride (13) on human melanoma cells (MEL-85, SK-MEL, A2058 and MEWO) [37]. The compound was found to decrease the viability and proliferation of MEL-85, SK-MEL, A2058 and MEWO cell lines *in vitro*, showing considerable cytotoxic activity on these human cells. Besides its cytotoxic properties, the compound was also found to interfere in cell features involved in tumour development such as adhesion to ECM components. This study proved the compound to be a potent drug against B16-F10 murine melanoma both *in vitro* and *in vivo*.

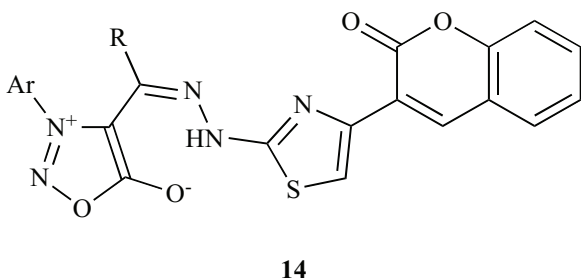


### Free Radical Scavenging

Stelia Carolina Mendez-Sanchez *et al.*, have studied the effect of 4-phenyl-5-(4-nitro-cinnamoyl)-1,3,4-thiadiazolium-2-phenylamine chloride (13) on lipoperoxidation and its ability to scavenge free radicals [38]. It was observed that the compound promoted a strong inhibition of the lipoperoxidation induced by  $\text{Fe}^{3+}$ -ADP/2-oxoglutarate in isolated mitochondria ( $95\% \pm 0.27$  at the highest concentration of  $80 \text{ nmol mg}^{-1}$  protein) in a dose-dependent manner. However, at the same concentration its effect was less intense ( $22\% \pm 3.46$ ) when the lipoperoxidation was initiated by peroxy radicals generated from the azocompound, AAPH. Lipid peroxidation in both coupled and uncoupled submitochondrial particles initiated with  $\text{Fe}^{2+}$ /NADH was also inhibited by the compound. The inhibition was about four times greater in coupled particles ( $34\%$  at  $80 \text{ nmol mg}^{-1}$  protein) in relation to uncoupled particles. The mesoionic compound showed ability of scavenging superoxide radical (7, 11 and 31% for 25, 38 and  $80 \mu\text{M}$ , respectively).

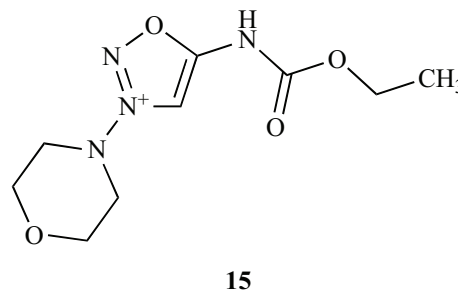
### Antidiabetic

Gireesh Tegginamath *et al.*, have reported a novel one-pot synthesis of sydnones appended to coumarins (14) via thiazole in the presence of silica sulphuric acid as a heterogeneous catalyst [39]. The compounds were screened for  $\alpha$ -amylase inhibition (antidiabetic) as well as DNA cleavage activity. The preliminary *in vitro* antidiabetic activity of these novel series of thiazole derivatives have shown that some of the chlorine substituted (electron withdrawing group) compounds exhibit good  $\alpha$ -amylase inhibition and good DNA cleavage activity.

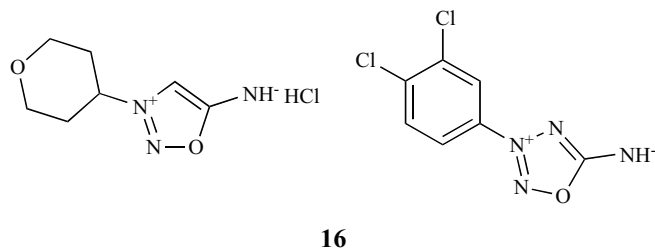


### Nitric Oxide Donor

Molsydmine (15) is an effective antihypertensive. This activity has been connected recently to its ability to release nitrogen monoxide *in vivo* [40]. This ability to act as a NO donor arises from hydrolysis of the side chain of molsydmine accompanied by decarboxylation and opening of the sydnonimine ring under these conditions to give *N*-nitroso derivative. It is known that *N*-nitroso compounds are readily oxidized in the organism, the final result of which is liberation of nitrogen monoxide.



Kankaanranta *et al.*, have studied the NO releasing capacity of two 3-aryl substituted oxatriazole-5-imine derivatives (16) and these were compared to a known NO donor, 4-Phenyl-5-(4-nitro-cinnamoyl)-1,3,4-thiadiazolium-2-phenylamine chloride [41]. It was found that both these derivatives induced an increase in cyclic GMP production in human platelets and neutrophils. The  $\text{IC}_{50}$  values were not found to be very different for both the derivatives. The authors have tried to establish a relationship between their biological activities and their ability to release nitric oxide.

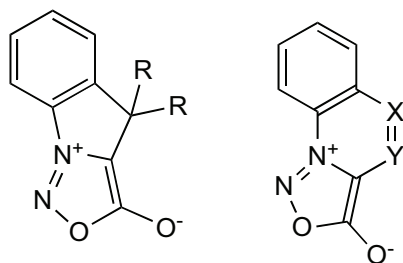


### FUTURE PERSPECTIVES

Sydnones and sydnonimines have been widely studied due to the array of biological activities exhibited by their nucleus. Studies have been carried out on sydnones with various substituents at the 4<sup>th</sup> position of the ring by subjecting them to a range of reactions like lithiation, bromination, modification of C4 halogenated sydnones. The cycloaddition of sydnones with various alkenes and alkynes have also gained significant interest in the recent years. Sydnones have been known to be novel precursors of pyrazoles (via cycloaddition). Some vasodilators like molsydmine and Darsydmine are known for their nitric oxide donor capacity.

Though 4-substituted sydnones have been fairly studied, fused ring sydnones have gained a lot of interest in recent years because ring fusion is a route which may lead to molecules with interesting biological activity. Mainly two types fused ring sydnones have been of interest (17).





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Very few reports are available on the synthesis and biological evaluation of these fused ring sydnones and their applications. Future work should, therefore, give importance to fused ring sydnones and their derivatives.

## CONCLUSION

In conclusion sydnones are highly versatile and robust members of the class of mesoionic compounds. They are known for their wide array of synthetic applications and biological activity. The literature reveals that sydnones have diverse biological potential. The easy synthetic routes to sydnones have attracted the attention of chemists, pharmacologists and researchers. Sydnones seem to have a great potential and remain largely unexplored.

## CONFLICT OF INTEREST

The author(s) confirm that this article content has no conflicts of interest.

## ACKNOWLEDGEMENTS

Declared none.

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